

Al-Fe-Mo (Aluminum-Iron-Molybdenum)

V. Raghavan

The early review of this system by [1992Rag] presented two isothermal sections at 1050 and 800 °C from the work of [1970Mar]. An update by [2005Rag] reviewed the isothermal section at 1000 °C determined by [1999Eum]. Extending this work, [2008Eum1, 2008Eum2] made a detailed investigation of the isothermal sections at 1150, 1000, and 800 °C.

Binary Systems

The Al-Fe phase diagram [1993Kat] shows a gamma loop. The body centered cubic (bcc) solid solution α exists in the disordered $A2$ form, as well as the ordered $B2$ and $D0_3$ forms. Apart from the high temperature phase ε ($D8_2$, Cu_5Zn_8 -type cubic; stable between 1232 and 1102 °C), there are three intermediate phases in the system: $FeAl_2$ (triclinic, space group $P\bar{1}$), Fe_2Al_5 (orthorhombic, space group $Cmcm$), and Fe_4Al_{13} (or $FeAl_3$; monoclinic, space group $C2/m$). The Al-Mo phase diagram [2008Eum1] depicts the following intermediate phases: $MoAl_{12}$

($Al_{12}W$ -type cubic), $MoAl_5$ (r) (rhombohedral, space group $R\bar{3}c$), $MoAl_5$ (h') (hexagonal, $P321$), $MoAl_5$ (h) (hexagonal, $P6_322$), Mo_5Al_{22} (orthorhombic, $Fdd2$), Mo_4Al_{17} (monoclinic, $C2$), $MoAl_4$ (monoclinic, Cm), $MoAl_3$ (monoclinic, $C2/m$), Mo_3Al_8 (monoclinic, Cm), Al_6Mo_3 , $MoAl$ (bcc), and Mo_3Al ($A15$, Cr_3Si -type cubic). In the Fe-Mo phase diagram [Massalski2], the σ phase ($D8_b$ -type tetragonal) and the R phase (rhombohedral) form through peritectic reactions and decompose above 1200 °C. Fe_7Mo_6 (denoted μ ; $D8_5$ -type rhombohedral) and Fe_2Mo ($C14$, $MgZn_2$ -type hexagonal) form in the solid state. The iron-based bcc phase α and (Mo) show significant mutual solubility.

Ternary Phases

Two ternary phases are known in this system: Al_8FeMo_3 (τ_1) has the $D0_{22}$, $\alpha TiAl_3$ -type of tetragonal structure with lattice parameters of $a = 0.3745$ nm and $c = 0.8390$ nm. It forms at a temperature above 1150 °C and decomposes eutectoidally at ~ 850 °C. It has a small homogeneity range

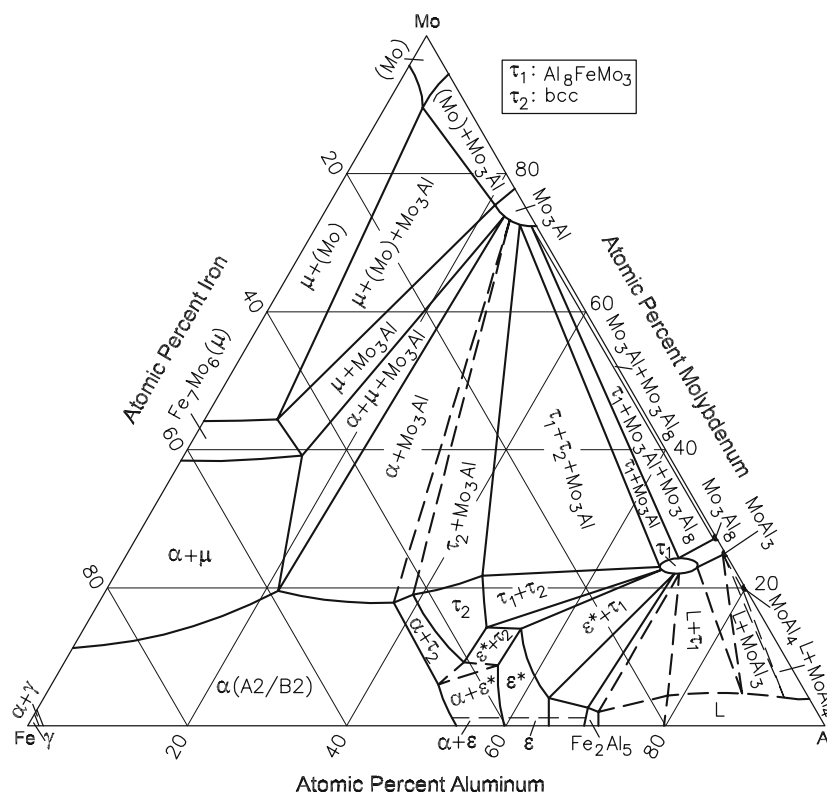


Fig. 1 Al-Fe-Mo isothermal section at 1150 °C [2008Eum2]

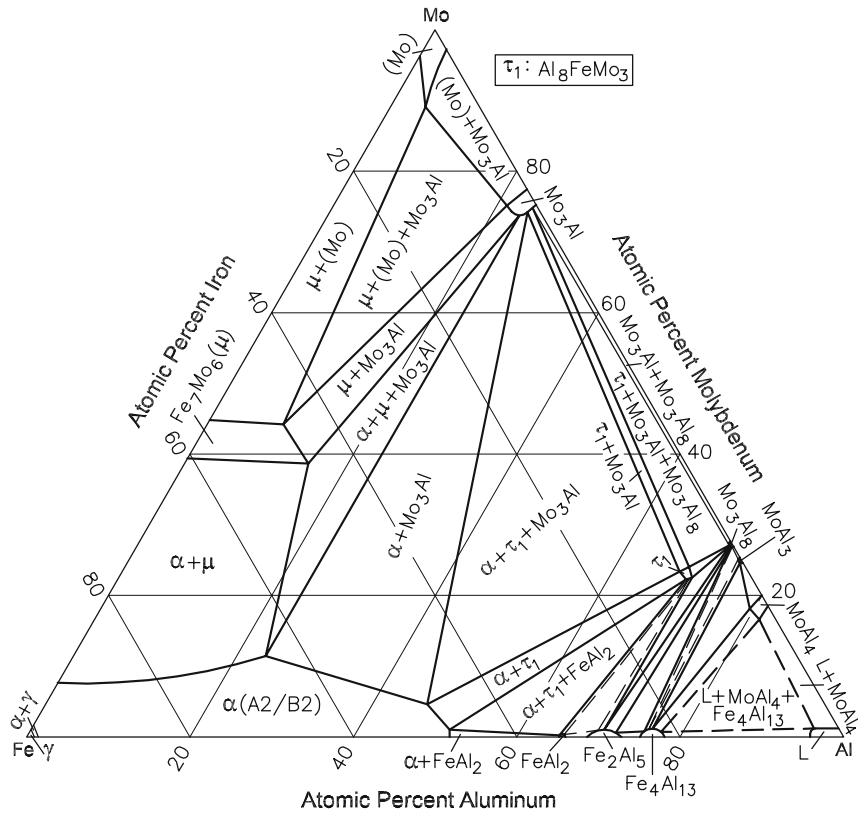


Fig. 2 Al-Fe-Mo isothermal section at 1000 °C [2008Eum2, 1999Eum]

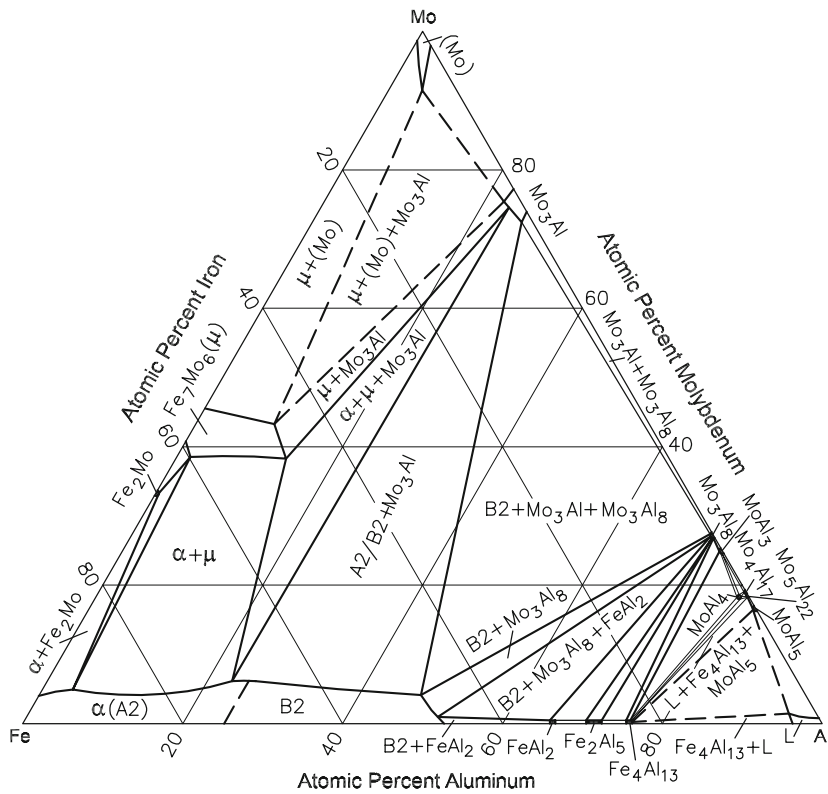


Fig. 3 Al-Fe-Mo isothermal section at 800 °C [2008Eum1]

Section II: Phase Diagram Evaluations

of 67.9-69.6 at.% Al and 22.0-24.2 at.% Mo at 1000 °C and 67.7-71.6 at.% Al and 22.1-24.4 at.% Mo at 1150 °C [2008Eum2]. The other phase τ_2 also forms above 1150 °C, has a homogeneity range and decomposes eutectoidally between 1071 and 1095 °C. It has the *A2*-type bcc structure and a lattice parameter range of 0.296-0.313 nm. On addition of Mo, the high-temperature Fe-Al phase ε , which possesses a $D8_2$, Cu_5Zn_8 -type cubic structure, undergoes a transition to ε^* with $D8_{10}$, Cr_5Al_8 -type rhombohedral structure [2008Eum2].

With starting metals of 99.999% Al, 99.95% Fe and 99.95% Mo, [2008Eum1, 2008Eum2] melted 13 binary and 54 ternary alloys by levitation melting under Ar atm. The alloys were annealed at 1150, 1000 and 800 °C for 200, 500 and 1000 h respectively and were quenched in iced brine. The phase equilibria were studied with the diffusion couple technique, optical and scanning/transmission electron microscopy, electron probe microanalysis, x-ray powder diffraction and differential thermal analysis. The isothermal sections constructed by [2008Eum2] at 1150, 1000 and 800 °C are shown in Fig. 1-3. At 1150 °C (Fig. 1), both ternary phases τ_1 and τ_2 are present. The τ_2 phase has an extended homogeneity range with the concentration of the three elements varying about 10 at.% each. The transition in the ε phase to the ε^* modification is indicated by a dotted line inside this phase region, which extends up to ~ 14 at.% Mo. The binary phases $FeAl_2$ and Fe_4Al_{13} were not detected by [2008Eum2] at 1150 °C.

At 1000 °C (Fig. 2) [2008Eum2, 1999Eum], τ_2 and $\varepsilon/\varepsilon^*$ phases are not stable and only τ_1 is present. At 800 °C (Fig. 3) [2008Eum1], none of the ternary phases are present.

The binary phase Al_4Mo , stabilized by a small amount of Fe, is present in the ternary region at 800 °C.

References

- 1970Mar:** V.Ya. Markiv, V.V. Burnashova, and V.R. Ryabov, Study of the Aluminum-Rich Part of the Phase Diagram of the Mo-Fe-Al System, *Dopov. Akad. Nauk Ukrain. RSR (A)*, 1970, (1), 69-72, in Ukrainian
- 1992Rag:** V. Raghavan, The Al-Fe-Mo (Aluminum-Iron-Molybdenum) System, *Phase Diagrams of Ternary Iron Alloys, Part 6A*, Indian Inst. Metals, Calcutta, 1992, p 135-141
- 1993Kat:** U.R. Kattner and B.P. Burton, Al-Fe (Aluminum-Iron), *Phase Diagrams of Binary Iron Alloys*, H. Okamoto, Ed., ASM International, Materials Park, OH, 1993, p 12-28
- 1999Eum:** M. Eumann, M. Palm, and G. Sauthoff, Constitution, Microstructure and Mechanical Properties of Ternary Fe-Al-Mo Alloys, *Euromat 99*, Vol. 10, D.G. Morris, S. Naka, and P. Caron, Ed., Wiley-VCH Verlag GmbH, Weinheim, Germany, 1999, p 146-153
- 2005Rag:** V. Raghavan, Al-Fe-Mo (Aluminum-Iron-Molybdenum), *J. Phase Equilib. Diffus.*, 2005, **26**(1), p 68-69
- 2006Eum:** M. Eumann, G. Sauthoff, and M. Palm, Re-evaluation of Phase Equilibria in the Al-Mo System, *Int. J. Mater. Res.*, 2006, **97**(11), p 1502-1511
- 2008Eum1:** M. Eumann, G. Sauthoff, and M. Palm, Phase Equilibria in the Fe-Al-Mo System. Part I: Stability of the Laves Phase Fe_2Mo and Isothermal Section at 800 °C, *Intermetallics*, 2008, **16**, p 706-716
- 2008Eum2:** M. Eumann, G. Sauthoff, and M. Palm, Phase Equilibria in the Fe-Al-Mo System. Part II: Isothermal Sections at 1000 and 1150 °C, *Intermetallics*, 2008, **16**, p 834-846